

## **Daily Tutorial Sheet-11**

**Numerical Value Type for JEE Main** 

- **126.(3)**  $CH_3COOH, CH_3COCH_3, CH_3CH(OH)CH_2CH_3$  will give haloform test
- 127.(4) aldehyde, formic acid, alpha hydroxy ketone will form Ag mirror with Tollens's reagent.
- 128.(1) aldehyde except benzaldehyde give Fehling's test.

Carbonyl carbon having + 1 oxidation state.

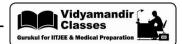
- **130.(2)** Aldehyde, methyl ketone and cyclic ketone give  $NaHSO_3$  test.
- **131.(7)** aldehyde, ketone, alcohol, 2° amine will react with 1 mole of RMgX each and ester will react with 2 mole of RMgX.

- **134.(4)** all aldehyde and ketonic group give 2, 4 DNP test but not glucose because it undergo cyclisation and exist in hemiacetal form.
- 135.(3) aldehyde are more reactive than ketone for nucleophilic addition reaction.

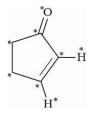
## 136.(11)

Sum of x + y is 11

**137.(4)** Compounds having alpha hydrogen will give aldol reaction.



**138.(8)** The molecule (A) is



No. of atoms lying in the same plane are 8

- **139.(2)** Compounds having no alpha hydrogen will give Cannizzaro reaction.
- **140.(3)** Rate expression for Cannizzaro reaction is Rate =  $[RCHO]^2[OH]$ So, the order of reaction is 3.